- 10. (a) (i) Explain with suitable examples of reducible and irreducible representation. (5)
 - (ii) Write a notes on direct products representation. (5)

Or

(b) Construct the character table for C₂V point group and discuss the uses of character table.

S.No. 352

17PCH03

(For the candidates admitted from 2017–2018 onwards)

M.Sc. DEGREE EXAMINATION, APRIL/MAY 2018.

First Semester

Chemistry

PHYSICAL CHEMISTRY — I

Time: Three hours

Maximum: 75 marks

PART A — $(5 \times 5 = 25 \text{ marks})$

Answer ALL questions.

1. (a) Explain C_p – C_v in terms of coefficient of expansion.

Or

- (b) Define the term partial molar volume.
- 2. (a) Discuss the thermodynamics of ideal and real gases.

Or

(b) Describe and explain mean activity coefficient.

3. (a) Discuss the hard sphere collision theory.

Or

- (b) Explain the influence of solvent on reaction in solutions.
- 4. (a) State and explain photo electric effect.

Or

- (b) Derive Schrodinger equation and its solution to particle present in 3-D box.
- 5. (a) Explain with suitable examples of

(i) Proper rotation (C_p)

(3)

(ii) Inversion centre (i).

(2)

Or

(b) State and explain the great orthogonality theorem and its consequences.

PART B — $(5 \times 10 = 50 \text{ marks})$

Answer ALL questions.

6. (a) Discuss the applications in the evaluation of $C_p - C_v$ for solids and for Vanderwaals gases.

Or

- (b) (i) Derive Gibbs-Duhem equation. (5)
 - (ii) How will you determine chemical potential by direct method? (5)

7. (a) What is Fugacity? Give the method of determination Fugacity and its variation of with temperature and pressure.

Or

- (b) How will you determine the activity coefficient from freezing point, EMF and solubility measurements?
- 8. (a) Explain with suitable mechanism of Lindemann theory of unimolecular reaction.

Or

- (b) (i) Explain with suitable examples of primary salt effect. (6)
 - (ii) Write a notes on kinetic isotopic effect.

(4)

- 9. (a) (i) Give the postulates of quantum mechanics. (7)
 - (ii) Write a notes on compton effect. (3)

Or

3

(b) Derive Schrodinger wave equation to harmonic oscillator. Calculate the energy of harmonic oscillator.