10. (a) Draw the correlation and FMO diagram for  $4\pi + 2\pi$  cycloaddition reaction. Explain.

Or

- (b) Discuss the following rearrangement.
  - (i) Sigmatropic.
  - (ii) Claisen.

(5 + 5)

S.No. 349

**12PCHZ04** 

(For the candidates admitted from 2012-2013 onwards)

M.Sc. DEGREE EXAMINATION, APRIL/MAY 2018.

Fourth Semester

Chemistry

ORGANIC SPECTRA, PHOTO CHEMISTRY AND PERICYCLIC REACTIONS

Time: Three hours

Maximum: 75 marks

PART A —  $(5 \times 5 = 25 \text{ marks})$ 

Answer ALL the questions.

1. (a) Describe the Woodward-Filser rules for calculating the  $\lambda_{max}$  value in enones.

Or

- (b) What are the factors affecting the vibrational frequencies in IR spectra? Explain.
- 2. (a) Explain axial haloketone rule.

Or and all visions

(b) Describe McLafferty rearrangement.

		보다 이 사람들이 가는 경기에서 하면 하는 사람들이 하는 것이다. 그렇게 하는 사람들이 하는 것이다. 그리를 하는 것이다.				
3.	(a)	Explain the principles of NMR spectroscopy.	7.	(a)	Explain:	
		Or			(i) Metastable ion peak. (3)	
	(b)	Write the applications of ESR spectroscopy.			(ii) Nitrogen rule. (4)	
1.	(a)	Draw and explain the Jablonski diagram.			(iii) Base peak. (3)	
		Or			Or	
	(b)	Explain Paterno-Buchi reaction.		(b)	Explain octant rule for ketones and + and -	
<b>5</b> .	(a)	Give an account on Woodward-Hofmann			cotton effect curves.	
		rule.	8.	(a)	Explain the various factors affecting	
		Or			chemical shift value.	
	(b)	Explain Cope rearrangement.			Or	
		PART B — $(5 \times 10 = 50 \text{ marks})$		(b)	Write notes on:	
		Answer ALL questions.			(i) Zero field splitting (5)	
6.	(a)	Discuss about the chromophores effect of			(ii) 'g' value. (5)	
		conjugation and substitutions with unshared electrons on $\lambda_{max}$ values for organic	9.	(a)	Explain Norrish type I and II reactions.	
		molecules.			Or	
		Or		(b)	Describe the following:	
	(b)	(i) Differentiate inter and intra molecular hydrogen bonding using IR spectra. (5)	,		(i) Photo-fries rearrangement.	
		(;) W;			(ii) Photo oxidation. (5 + 5)	â

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