9. (a) What are Ferrocenes? Explain the structure and bonding in Ferrocenes.

Or

- (b) Discuss the following reaction of organometallic compounds with mechanism
 - (i) oxidative addition
 - (ii) reductive elimination. (5+5)
- 10. (a) (i) What is Catalyst? Give the general characteristics of a catalyst.
 - (ii) Explain the mechanism of hydroformylation of olefins using Rhodium catalyst. (5+5)

Or

- (b) Describe the mechanism of:
 - (i) Wilkinson's catalyst
 - (ii) Zeigler Natta catalyst. (5+5)

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12PCHZ02/ 12POCZ02

(For the candidates admitted from 2012-2013 onwards)

M.Sc. DEGREE EXAMINATION, APRIL/MAY 2018.

Second Semester

Chemistry

COORDINATION CHEMISTRY

(Common for Organic Chemistry)

Time: Three hours

Maximum: 75 marks

PART A — $(5 \times 5 = 25 \text{ marks})$

Answer ALL questions.

1. (a) State Jahn-Teller theorem. Apply the theorem to d⁸ and d⁹ systems.

Or

- (b) Discuss the factors affecting splitting orbitals.
- 2. (a) Explain stepwise and overall formation constants.

Or

(b) Bring out the application of CD in the identification of Chirality of complexes.

3.	(a)	Explain the following:	
		(i)	Cross reaction
		(ii)	Marcus-Hush theory.

 $(2\frac{1}{2} + 2\frac{1}{2})$

Or

- (b) Discuss the hydrolysis of acid and base mechanism.
- 4. (a) Bring out the synthesis and structure of acetylene complexes.

Or

- (b) What is Organometallics? Explain with example.
- 5. (a) Discuss the mechanism of Oxo process.

Or

(b) Describe the mechanism of Wacker process.

PART B — $(5 \times 10 = 50 \text{ marks})$

Answer ALL questions.

- 6. (a) Define the following:
 - (i) MO theory
 - (ii) Energy level diagram. (5+5)

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- (b) Discuss:
 - (i) Sigma and Pi bonding in complexes
 - (ii) Magnetic properties of complexes.
- 7. (a) What is meant by stability constant? On what factors does the stability of a complex ion depend? How stability constant of a given complex is determined by Spectro photometrically?

Or

- (b) Explain the structure of
 - (i) Schiff's base
 - (ii) Cryptates.

(5+5)

- 8. (a) Discuss the following:
 - (i) Labile and inert complex
 - (ii) Substitution in square planar complexes. (5+5)

Or

(b) Explain the experimental evidences in favour of SNCB mechanism.