- 10. (a) (i) Discuss the principle of Harmonic oscillator of vibrational spectra. (7)
 - (ii) Write a brief account on fermi resonance. (3)

Or

- (b) Write a notes on the following
 - (i) Hot bands. (2)
 - (ii) Over tones and combination frequencies. (4)
 - (iii) Electronic spectra of diatomic molecule.

(4)

S.No. 257

12PCH03/ 12PAC03

(For the candidates admitted from 2012-2013 onwards)

M.Sc. DEGREE EXAMINATION, NOVEMBER 2017.

First Semester

Chemistry

PHYSICAL CHEMISTRY - I

(Common for M.Sc. Analytical Chemistry)

Time: Three hours

Maximum: 75 marks

PART A — $(5 \times 5 = 25 \text{ marks})$

Answer ALL questions.

1. (a) Derive the relation between cp-cv.

Or

- (b) Define the term chemical potential.
- 2. (a) State and explain arrhenius theory. Discuss its significances.

Or

(b) Give the comparison between collision theory and Activated complex theory.

3.	(a)	Write a notes on photo electric effect.	7.	(a)	Explain with suitable mechanism of Hard- sphere collision theory of reaction rate.
		Or			Or
	(b)	Discuss the postulates of quantum mechanics.		(b)	Give the method of determination of the following:
4.	(a)	Explain with suitable example of proper rotation.			(i) Free energy (4)
		Or			(ii) Enthalpy (3)
	(b)	Define and explain direct product			(iii) Entropy of Activation. (3)
	(6)	representation.	8.	(a)	Write a brief account on
5.	(a)	State and explain Einstein's theory of			(i) Heisenberg's uncertainty principle. (4)
		transistion probability.			(ii) Hamiltonion operator. (6)
		Or			Or
	(b)	State and explain Frank-Condon principle.		(b)	Derive schrodinger equation and its solution
		PART B — $(5 \times 10 = 50 \text{ marks})$			to the problem of a particle in Hormonic oscillator.
		Answer ALL questions.	9.	(a)	(i) Explain with suitable examples of
6.	(a)	(i) Give the Maxwell's relations and thermodynamic equations of state. (6)	3.	(a)	symmetry elements and symmetry operation. (7)
		(ii) Discuss the applications of in the evaluation of cp-cv for solids. (4)			(ii) What is point group? Explain with suitable examples. (3)
		Or			Or
	(b)	(i) How will you determine the chemical potential by direct method. (5)		(b)	(i) State and explain the great orthogonality theorem. and its
		(ii) How the chemical potential varies with pressure. Explain. (5)			consequences. (7) (ii) Define the term reducible and irreducible representation. (3)
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